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Synthesis of 2-amino-3-hydroxy-4-substituted pyridines via regioselective metalation of 3-(1-ethylpropyl)-[1,3]oxazolo[4,5-b]pyridin-2(3H)-one and application to corticotropin releasing factor₁ receptor ligands

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Abstract—An efficient route to the preparation of 2-amino-3-hydroxy-4-substituted pyridines is described. The key step involves the regioselective metalation and subsequent alkylation of the [1,3]oxazolo[4,5-b]pyridin-2(3H)-one ring system. Base hydrolysis provides access to a variety of 4-substituted pyridines. This chemistry is proved to be useful for the synthesis of corticotropin releasing factor₁ receptor ligands.

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During the course of our investigation to identify novel ligands for corticotropin releasing factor₁ (CRF₁) receptors¹ we sought an efficient method for the preparation of 4-substituted pyridines. We were particularly interested in a synthetic route that would be readily amenable to the preparation of compounds bearing a phenyl substituent at the 4-position of the pyridine ring.

Directed *ortho* metalation has been widely used for the regioselective alkylation and arylation of aryl and heteroaryl ring systems.² Our synthetic plan began with commercially available 2-amino-3-hydroxy pyridine. It was anticipated that installation of the substituent at the 4-position could be carried out by a directed *ortho* metalation reaction whereby the oxygen at the 3-position would serve as a directing group. This plan was bolstered by the previously reported regioselective metalation of related heterocyclic ring systems.^{3,4} The 2-amino and 3-hydroxy groups could be protected as an oxazolidinone, which would be ideal since it could

be easily hydrolyzed to regenerate the amino and hydroxy groups later in the synthetic route.

The synthesis begins with alkylation of the amino group of 2-amino-3-hydroxy pyridine (1) with 3-pentanone via reductive amination (Scheme 1). Subsequent treatment with triphosgene resulted in the formation of the [1,3]oxazolo[4,5-b]pyridin-2(3H)-one ring system (3). An attempt to directly couple 3 with bromobenzene via a Negishi coupling reaction resulted in a low yield of desired product (Table 1, compound 11). Regioselective metalation with t-BuLi resulted in the formation of a mixture of the desired lithium anion at the 4-position of the pyridine ring as well as addition of t-BuLi to the oxazolidinone carbonyl group resulting in the formation of 2,2-dimethylpropionic acid 2-(1-ethyl-propylamino)pyridin-3-yl ester 16 as the major product in 62% yield (Fig. 1).⁵ Cleavage of the C-N bond rather than the C-O bond is likely a result of the enhanced leaving group ability of this nitrogen due to resonance stabilization of the resulting charge into the neighboring pyridine nitrogen. Within several minutes of isolation, the product turned blue in color, which was likely due to trace amounts of air oxidation.

Formation of this undesired byproduct was circumvented by using a two-step procedure to install the

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Scheme 1. Reagents and conditions: (a) 3-pentanone, NaBH₃CN, THF, 62%; (b) triphosgene, Et₃N, THF 70%; (c) LDA, BrCH₂CH₂Br, THF, 65%; (d) Ba(OH)₂·8H₂O, Pd(PPh₃)₂Cl₂, 2,4-dichlorobenzeneboronic acid, DME, H₂O, 80%; (e) KOH, EtOH, 87%.

Table 1.

Compds	Electrophile	R	Yield (%)
4	BrCH ₂ CH ₂ Br	Br	65
7	MeI	Me	78
8	СНО	V OH	89
9	CHO	✓ \	73
10	N.	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	76
11 ^a	PhBr	Ph	11

^a Compound 11 was prepared by a Negishi coupling reaction (*t*-BuLi, ZnCl₂, Pd(PPh₃)₄, PhBr).

Figure 1. Structure of compound 16.

phenyl group. Deprotonation with LDA⁶ followed by the addition of dibromoethane³ cleanly afforded desired bromide 4 in 65% yield. Compound 4 was then subjected to Suzuki coupling conditions to furnish biaryl intermediate 5 in good yield. Treatment with NaOH resulted in hydrolysis of the oxazolidinone to form a 2-amino-3-hydroxy-4-substituted pyridine (6).

Having successfully completed the synthesis of the desired 4-substituted pyridyl ring system (6) we were inter-

ested in studying the directed metalation of 3 in more detail. The intermediate lithium anion was subjected to additional electrophiles as shown in Table 1. Addition of methyl iodide, several aldehydes, and a Weinreb amide each proceeded in good yield. Not unexpectedly, treatment with ethyl iodide resulted only in recovery of protonated starting material due to competing elimination.

With 2-amino-3-hydroxy-4-substituted pyridine 6 in hand, it was elaborated further to the desired products (Scheme 2) which were tested for CRF₁ receptor binding affinity. Compound 6 was selectively alkylated on either the amine or hydroxyl groups depending upon the choice of reaction conditions. Subjection of 6 to formal-dehyde under reductive amination conditions resulted in the selective formation of 12; whereas, treatment of 6

Scheme 2. Reagents and conditions: (a) HCHO, ZnCl₂, NaBH₃CN, MeOH, 60 °C, 34%; (b) K₂CO₃, EtI, acetone, 81%; (c) K₂CO₃, methyl 2-bromopropionate, 2-butanone, 74%; (d) *p*-TsOH·H₂O (cat), toluene, 4 days, 96%; (e) BH₃·SMe₂, THF, 84%.

with K₂CO₃ and ethyl iodide resulted in the selective alkylation of the phenol oxygen to form **13**. It was also possible to prepare bicyclic compounds by taking advantage of the differential reactivity of the amine and phenol groups on the pyridine ring. Treatment of **6** with 2-bromomethylpropionate⁷ followed by heating in toluene in the presence of *p*-TsOH (4 days) resulted in the formation of **14**. The amide was then reduced with BH₃·SMe₂ to form **15** in good yield.⁸

Compounds 12, 13, and 15 were tested for their binding affinity to the CRF₁ receptor. Compounds 12 and 13 were found to be inactive ($K_i > 10,000 \text{ nM}$); however, compound 15 showed moderately potent binding affinity ($K_i = 111 \pm 10 \text{ nM}$, n = 4) for the CRF₁ receptor. Binding affinities were determined in a CRF₁ receptor binding titration assay using rat frontal cortex homogenate, in which inhibition of specific binding of [125 I]ovine–CRF by our test compounds was measured to determine their receptor binding affinity. 9,10

In conclusion, the preparation of 2-amino-3-hydroxy-4-substituted pyridines was carried out by the regioselective metalation of the [1,3]oxazolo[4,5-b]pyridin-2(3H)-one ring system (3) as a key step. Suzuki coupling followed by base hydrolysis furnished the 4-substituted pyridine intermediate, which was subsequently converted to the desired products. This method appears to be general, allowing ready access to a variety of 2-amino-3-hydroxy-4-substituted pyridines, and was also directly applicable to our investigation of CRF₁ receptor ligands.

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- 5. The structure of compound **16** was confirmed by a COSY experiment in addition to the following characterization data for this compound: blue oil; ¹H NMR (400 MHz, CDCl₃) δ 7.93 (dd, J = 5.1, 1.5 Hz, 1H), 7.17 (dd, J = 7.5, 1.5 Hz, 1H), 6.51 (dd, J = 7.8, 2.8 Hz, 1H), 4.11 (d, J = 8.1 Hz, 1H), 4.02–3.97 (m, 1H), 1.64–1.55 (m, 2H), 1.54–1.41 (m, 2H), 1.37 (s, 9H), 0.88 (t, J = 7.3 Hz, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 175.99, 151.39, 144.62, 132.96, 128.06, 111.37, 52.26, 39.37, 27.20, 26.95, 9.79; IR

- (thin film) 2965 (s), 2935 (m), 2875 (m), 1758 (s), 1609 (s), 1500 (s), 1169 (s), 1104 (s) cm $^{-1}$; HRMS (ESI) *mle* 265.1918 [(M+H) $^+$, calcd for $C_{15}H_{25}N_2O_2$ 265.1916].
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- 8. Characterization data for compounds 2-15 follows: compound 2: yellow solid; mp 112-113 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.52 (d, J = 5.5 Hz, 1H), 6.81 (d, J = 7.3 Hz, 1H), 6.41–6.37 (m, 1H), 4.95 (br, 1H), 3.94– 3.85 (m, 1H), 1.68-1.46 (m, 4H), 0.91 (t, J = 7.3 Hz, 6H);HRMS (ESI) m/e 181.1343 $[(M + H)^+, \text{ calcd for } C_{10}H_{17}N_2O$ 181.1341]; **3**: colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 8.09 (dd, J = 5.5, 1.5 Hz, 1H), 7.40 (dd, J = 7.7, 1.1 Hz, 1H), 7.03 (dd, J = 7.7, 5.2 Hz, 1H), 4.25-4.15 (m, 1H), 2.26-2.14 (m, 2H), 1.96-1.82 (m, 2H), 0.89 (t, J = 7.3 Hz, 6H); HRMS (ESI) m/e 207.1141 $[(M+H)^+, \text{ calcd for } C_{11}H_{15}N_2O_2 \text{ 207.1134}]; \text{ compound 4: colorless solid; mp 83.5–84.5 °C; }^1H NMR (300 MHz,$ CDCl₃) δ 7.91 (d, J = 5.5 Hz, 1H), 7.19 (d, J = 5.8 Hz, 1H), 4.21–4.12 (m, 1H), 2.24–2.11 (m, 2H), 1.95–1.81 (m, 2H), 0.89 (t, J = 7.7 Hz, 6H); HRMS (ESI) m/e 285.0236 $[(M+H)^{+}, calcd for C_{11}H_{14}N_{2}O_{2}Br 285.0239]; compound$ 5: blue-green solid; ${}^{1}H$ NMR (300 MHz, CDCl₃) δ 8.15 (d, J = 5.4 Hz, 1H), 7.57 (d, J = 1.4 Hz, 1H), 7.43–7.36 (m, 2H), 7.09 (d, J = 5.5 Hz, 1H), 4.26–4.18 (m, 1H), 2.29–2.16 (m, 2H), 1.99-1.85 (m, 2H), 0.93 (t, J = 7.3 Hz, 6H); LRMS (APCI) m/e 351.1 [(M+H)⁺, calcd for $C_{17}H_{17}$ -N₂O₂Cl₂ 351.1]; compound **6**: green amorphous solid; ¹H NMR (300 MHz, CDCl₃) δ 6.63 (d, J = 5.2 Hz, 1H), 7.53 (d, J = 2.1 Hz, 1H), 7.34 (dd, J = 8.4, 1.8 Hz, 1H), 7.28 (d, J = 8.1 Hz, 1H), 6.28 (d, J = 5.1 Hz, 1H), 4.88 (br, 1H), 3.98 (br, 1H), 1.71-1.49 (m, 4H), 0.94 (t, J = 7.3 Hz, 6H); HRMS (ESI) m/e 325.0888 [(M+H)⁺, calcd for C₁₆H₁₉N₂OCl₂ 325.0874]; compound 7: colorless solid; mp 65–66 °C, 1 H NMR (300 MHz, CDCl₃) δ 7.96 (d, J = 5.5 Hz, 1H), 6.86 (d, J = 5.8 Hz, 2H), 4.23-4.13 (m,1H), 2.39 (s, 3H), 2.28-2.13 (m, 2H), 1.94-1.80 (m, 2H), (t, J = 7.3 Hz, 6H); LRMS (APCI) m/e $221.1[(M+H)^{+}$, calcd for $C_{12}H_{17}N_{2}O_{2}$ 221.1]; compound 8: light yellow solid; mp 115.5–116.5 °C, ¹H NMR (300 MHz, CDCl₃) δ 8.08 (d, J = 5.5 Hz, 1H), 7.51 (dd, J = 8.4, 1.8 Hz, 2H), 7.42–7.31 (m, 3H), 7.24 (d, J = 5.5 Hz, 1H, 6.15 (s, 1H), 4.21-4.13 (m, 1H), 2.23-4.13 (m, 1H)2.09 (m, 3H), 1.92-1.83 (m, 2H), 0.88 (t, J = 7.7 Hz, 6H);HRMS (ESI) m/e 313.1543 $[(M+H)^+, calcd for$ $C_{18}H_{21}N_2O_3$ 313.1552]; **9**: pale blue oil; ¹H NMR (300 MHz, CDCl₃) δ 8.07 (d, J = 5.5 Hz, 1H), 7.14 (d, J = 5.5 Hz, 1H), 4.98 (t, J = 6.6 Hz, 1H), 4.23–4.13 (m, H), 2.53 (br, H), 2.23–2.11 (m, H), 1.94–1.80 (m, 4H), 0,99 (t, J = 7.3 Hz, 3H), 0.87 (t, J = 7.4 Hz, 6H); LRMS (APCI) m/e 265.0 $[(M+H)^+, \text{ calcd for } C_{14}H_{21}N_2O_3 265.2];$ compound 10: pale blue oil; ¹H NMR (300 MHz, CDCl₃) δ 8.17 (d, J = 5.5 Hz, 1H), 7.48 (d, J = 5.5 Hz, 1H), 4.28– 4.18 (m, 1H), 4.14-4.08 (m, 1H), 2.38 (q, J = 8.0 Hz, 4H),2.28-2.09 (m, 3H), 1.98-1.83 (m, 1H), 0.89 (t, J = 7.7 Hz, 6H); LRMS (ESI) m/e 289.1540 [(M+H)⁺, calcd for $C_{16}H_{21}N_2O_3$ 289.1540]; compound 11: blue oil; ¹H NMR $(300 \text{ MHz}, \text{ CDCl}_3) 8.15 \text{ (d, } J = 5.8 \text{ Hz}, \text{ 1H)}, 7.90 \text{ (dd,}$ J = 6.6, 2.1 Hz, 2H), 7.58–7.44 (m, 3H), 7.28 (d, J = 5.5 Hz, 1H, 4.29-4.20 (m, 1H), 2.38-2.20 (m, 2H),2.00-1.82 (m, 2H), 0.94 (t, J = 7.3 Hz, 6H); HRMS (ESI) m/e 283.1450 [(M+H)⁺, calcd for C₁₇H₁₉N₂O₂ 283.1447]; 12: light green solid; mp 72.6–73.2 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.98 (d, J = 5.1 Hz, 1H), 7.53 (d, J = 1.8 Hz, 1H), 7.36–7.28 (m, 2H), 6.87 (d, J = 5.1 Hz, 1H), 2.99-2.95 (m, 1H), 2.71 (s, 3H), 1.69-1.52 (m, 4H),

0.92 (t, J = 7.3 Hz, 6H); HRMS (ESI) mle 339.1024 [(M+H)⁺, calcd for $C_{17}H_{21}N_2OCl_2$ 339.1031]. Anal. Calcd for $C_{17}H_{20}N_2OCl_2$: C, 60.18; H, 5.94; N, 8.27. Found: C, 60.27; H, 5.90; N, 8.16; compound 13: dark blue oil; ¹H NMR (300 MHz, CDCl₃) δ 7.85 (d, J = 5.5 Hz, 1H), 7.51 (m, 1H), 7.30 (d, J = 1.1 Hz, 2H), 6.35 (d, J = 5.1 Hz, 1H), 4.85 (d, J = 9.2 Hz, 1H), 4.07–4.00 (m, 1H), 3.53 (q, J = 6.9 Hz, 2H), 1.72–1.49 (m, 4H), 1.10 (t, J = 7.0 Hz, 3H), 0.95 (t, J = 7.3 Hz, 6H); HRMS (ESI) mle 353.1216 [(M+H)⁺, calcd for $C_{18}H_{23}N_2OCl_2$ 353.1187]. Anal. Calcd for $C_{18}H_{22}N_2OCl_2$: C, 61.20; H, 6.29; N, 7.94. Found: C, 61.06; H, 6.31; N, 7.71; compound 14: dark oil; ¹H NMR (300 MHz, CDCl₃) δ 8.04 (d, J = 4.8 Hz, 1H), 7.52 (d, J = 2.1 Hz, 1H), 7.35 (m, 1H), 7.25 (d, J = 4 Hz,

1H), 5.1 (m, 1H), 4.66 (q, J = 6.5 Hz, 1H), 2.31 (m, 2H), 1.91 (sept, J = 7.3 Hz, 2H), 1.51 (d, J = 7.0 Hz, 1H), 0.89 (t, J = 7.4 Hz, 6H); HRMS (ESI) mle 379.0980 [(M+H)⁺, calcd for $C_{19}H_{21}Cl_2N_2O_2$ 379.2797]; compound **15**: colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 7.74 (d, J = 5.1 Hz, 1H), 7.47 (d, J = 2.2 Hz, 1H), 7.30 (m, 2H), 6.32 (d, J = 5.1 Hz, 1H), 4.86 (m, 1H), 4.15 (m, 1H), 3.30 (dd, J = 12.0, 2.2 Hz, 1H), 3.02 (m, 1H), 1.64 (m, 4H), 1.28 (d, J = 6.2 Hz, 3H), 0.93 (t, J = 7.3 Hz, 6H); HRMS (ESI) mle 365.1169 [(M+H)⁺, calcd for $C_{19}H_{23}N_2OCl_2$ 365.2962].

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